



Harnessing proton-conducting ceramics to produce methane and oxygen on Mars

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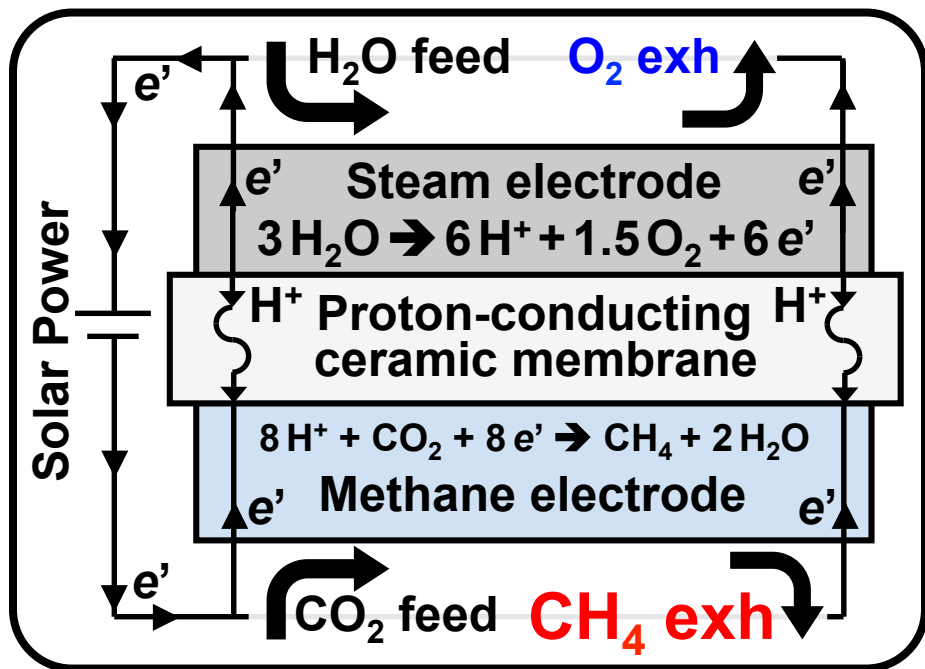
We are using proton-conducting ceramics to convert H_2O & CO_2 into CH_4 and O_2



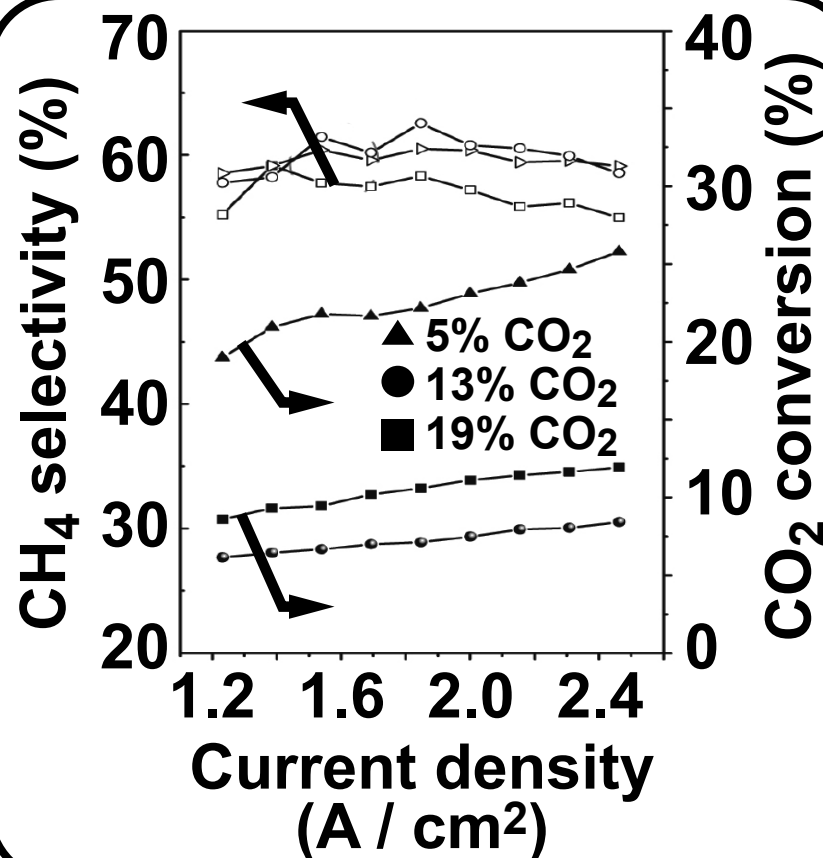
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Concept



Performance



We have been developing proton-conducting ceramics to convert natural gas into gasoline



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- First inspired by CoorsTek, Inc.
 - Dr. W. Grover Coors
 - Long history with Mines
- Methane dehydroaromatization (MDA)
 - Upgrade natural gas into benzene
$$6 \text{ CH}_4 \rightarrow \text{C}_6\text{H}_6 + 9 \text{ H}_2$$
 - Suffers from poor CH_4 conversion
 - Thermodynamic limitation
- Proton-conducting membrane
 - Remove H_2 from product stream
 - Shift thermo towards products
 - Increase CH_4 conversion
 - Produce more C_6H_6 and the like

COORSTEK

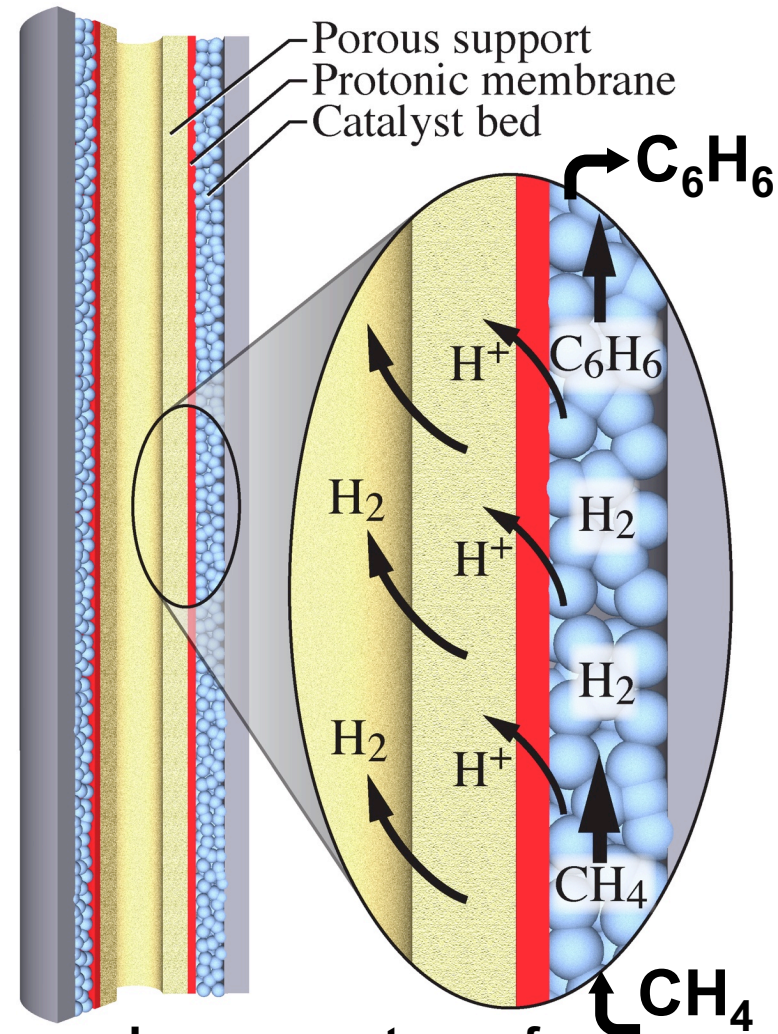


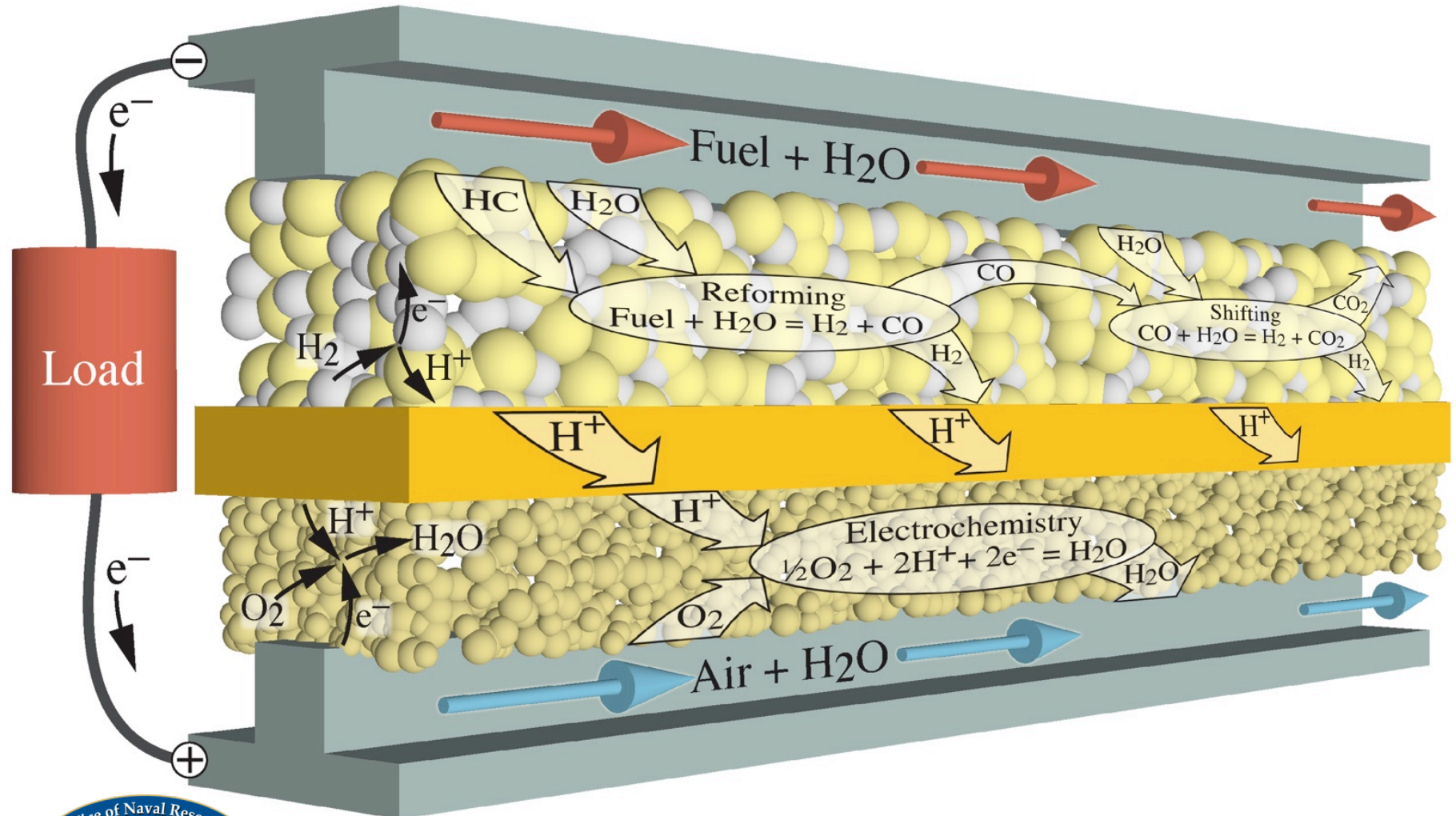
Image courtesy of
Prof. Robert J. Kee, CSM

MDA produced breakthroughs in fuel-cell applications for efficient, distributed electricity generation



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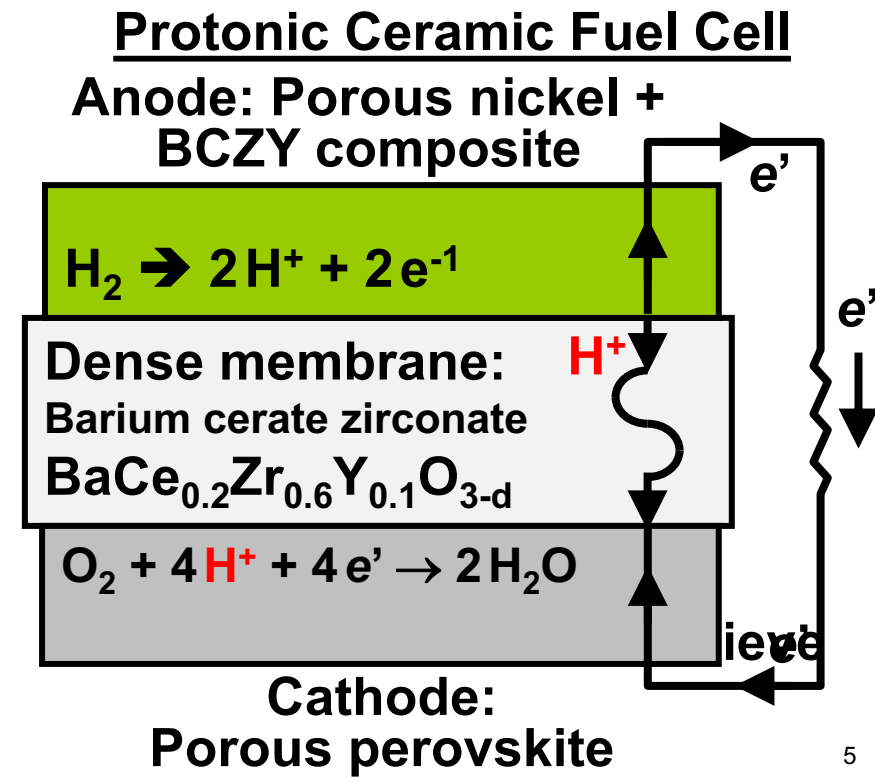
Proton-conducting ceramics are an emerging material with broad energy applications



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- Ceramic ion conductors are an active area of research
 - Yttria-stabilized zirconia (YSZ) is the current state-of-the-art
 - Oxygen ions (O^{2-}) serve as charge carriers
 - Require 700 – 850 °C operation
- Protonic ceramics have a number of performance benefits
 - Small charge carrier (H^+)
 - Higher ionic conductivity
 - Lower-temperature operation
 - 500 – 600 °C operation
 - Lower degradation rates
 - Lower cost balance of plant
- ABO_3 perovskite structure
 - $BaCe_xZr_yY_{1-x-y}O_{3-d}$
 - Tune stoichiometry to desired properties



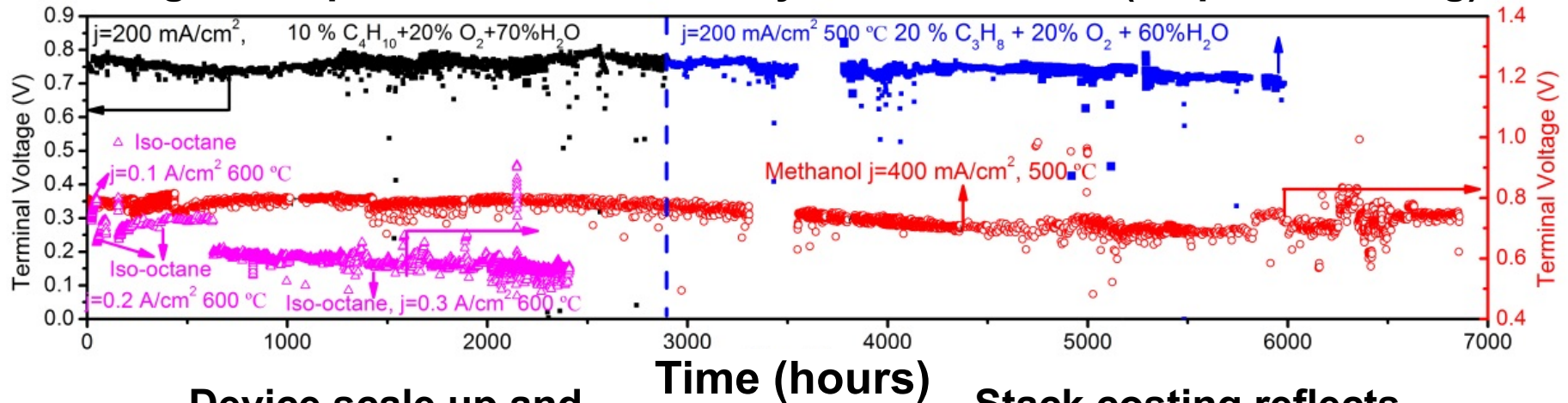
Our Phase I REBELS effort led to breakthroughs in intermediate-temperature fuel-cell performance



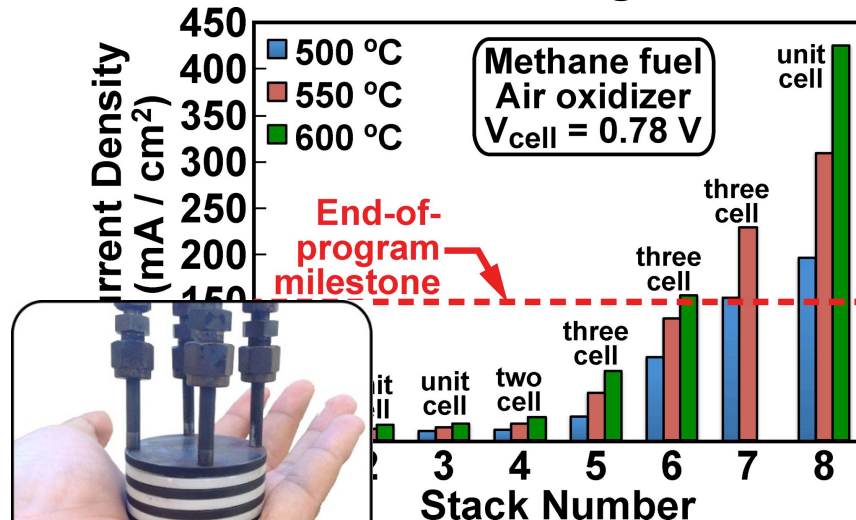
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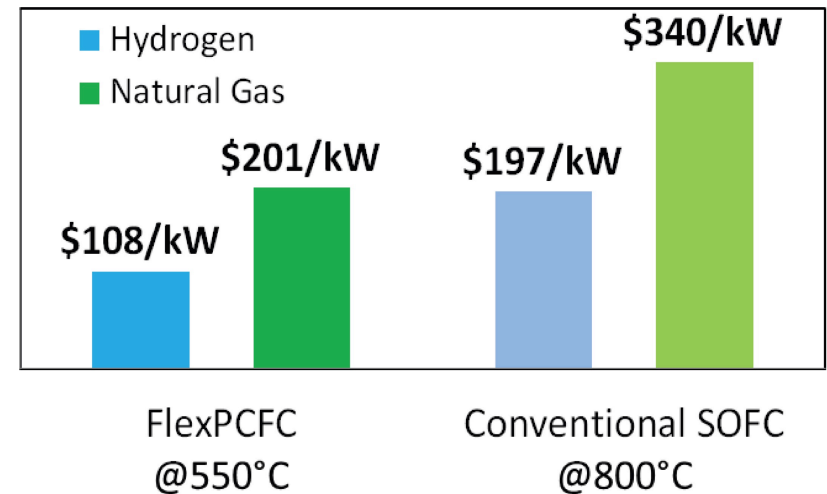
Long-term operation under direct hydrocarbon fuels (no pre-reforming)



Device scale up and multi-cell stack integration



Stack costing reflects protonic-ceramic advantages

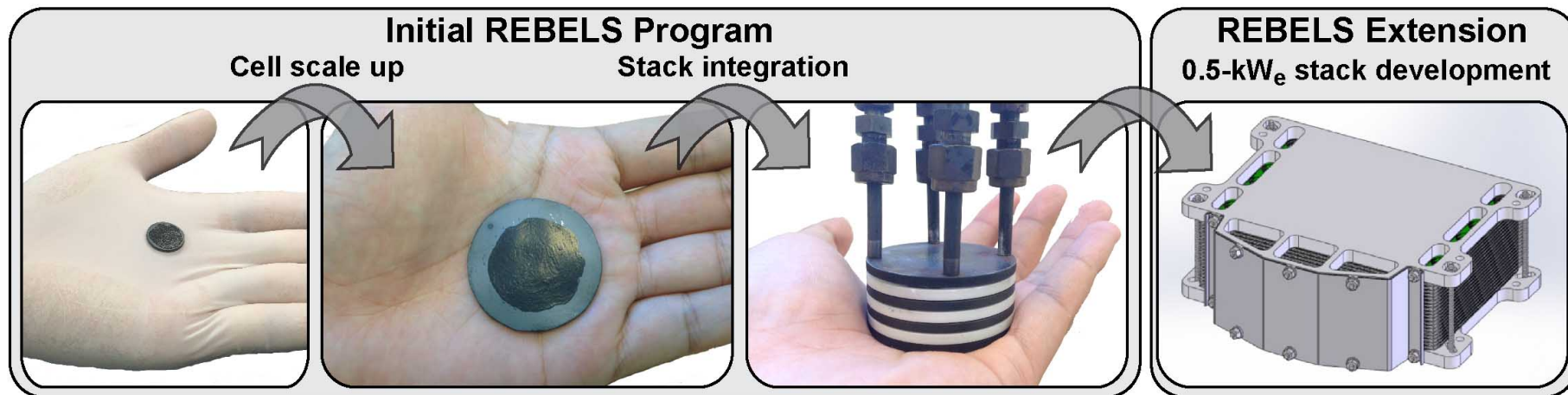


Promising results justify further protonic-ceramic fuel-cell scale up to the 500-W stack level in Phase II



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- New industrial partner: FuelCell Energy (Danbury, Connecticut, USA)
- Further scale up of cell size by order of magnitude
- Explore and mitigate degradation
- Optimize hydrocarbon internal reforming conditions
- Understand “value proposition” of protonic-ceramic fuel cells



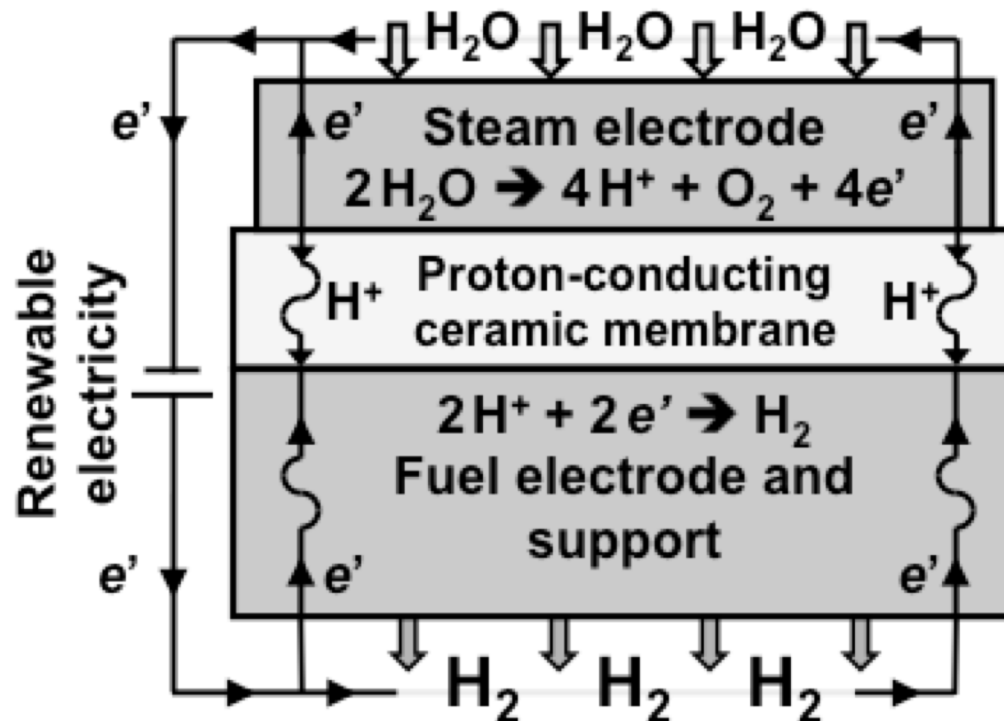
We are extending protonic ceramics for steam electrolysis, H₂ generation, and energy storage



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- Pure H₂ product stream is formed, unlike O²⁻ electrolyzers
- Eliminates costly downstream H₂-separation processes
- Serves to store intermittent renewable energy in form of hydrogen
- Recent award from DOE Fuel Cell Technologies Office
 - Two-year program; target start date of August 2018



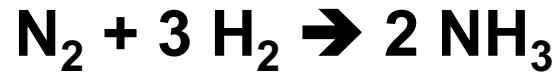
We are now extending protonic ceramics to chemicals synthesis for renewable-energy storage



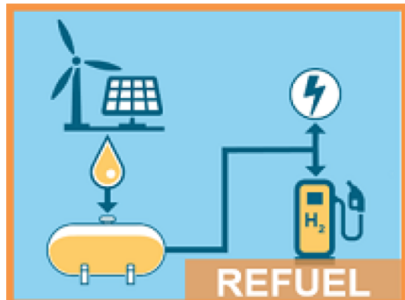
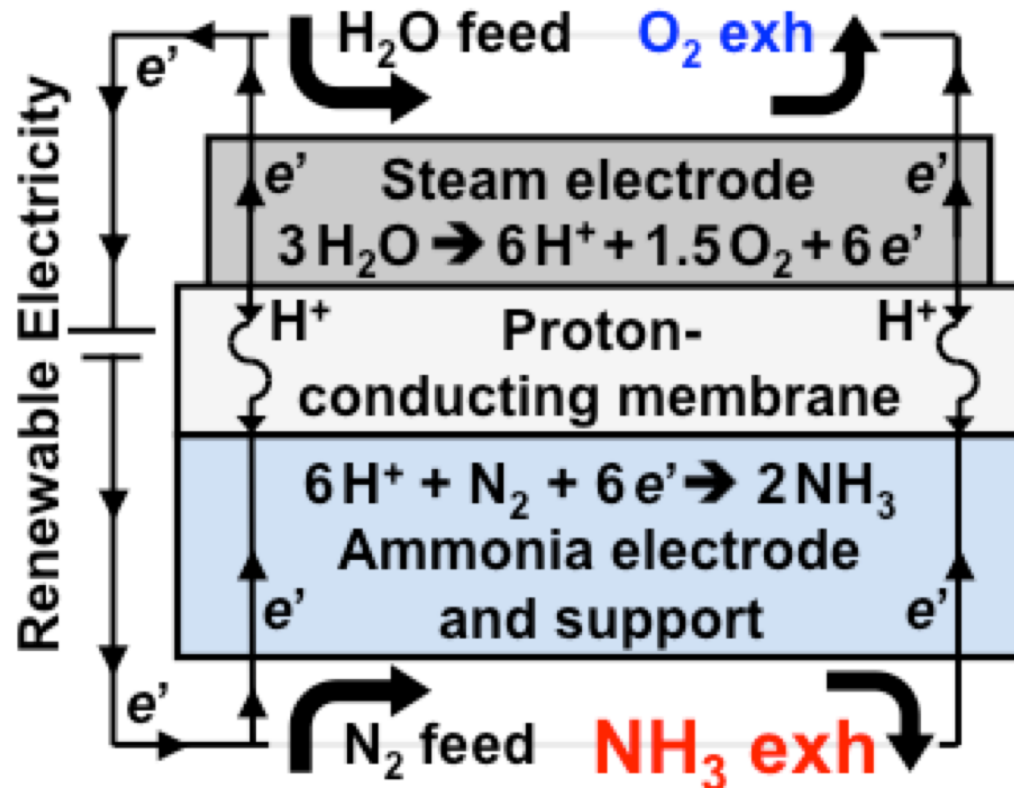
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- Ammonia presents an effective energy-storage solution



- Create carbon-neutral liquid fuel to store intermittent renewables

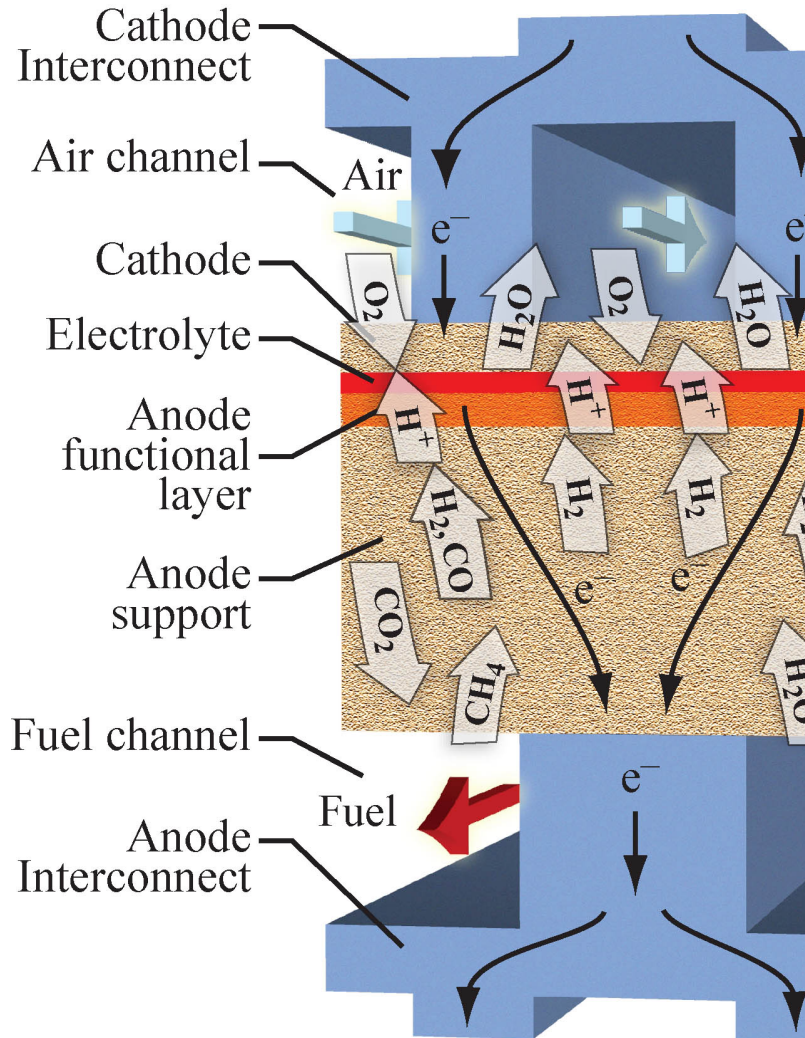


Central objective is operation under 100% internal reforming of natural gas fuel premixed with steam



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- Steam injected with CH₄ fuel
- H₂ consumption shifts chemistry
$$\text{CH}_4 + 2 \text{H}_2\text{O} \rightarrow 4 \text{H}_2 + \text{CO}_2$$
- Gas transport through porous anode
 - Optimal anode morphology
 - High porosity
 - Large pore diameters
 - Open microstructure
- Hydrocarbon internal reforming
 - Catalytic processes
 - Optimal anode morphology
 - High catalyst surface area
 - Lower porosity
 - Tight microstructure
- Conflicting anode-design objectives

We have a new program with NASA for harnessing protonic ceramics to make fuel on Mars



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- Convert CO_2 and H_2O into O_2 and CH_4



CO_2 and H_2O into O_2 and CH_4

Operation of ceramic electrochemical devices on Mars presents “balance-of-plant” challenges

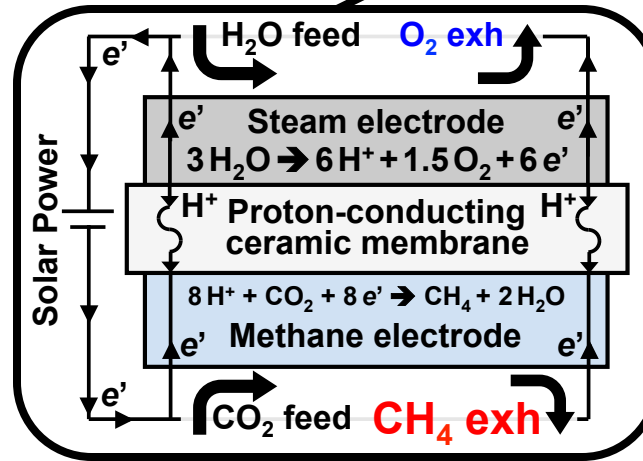
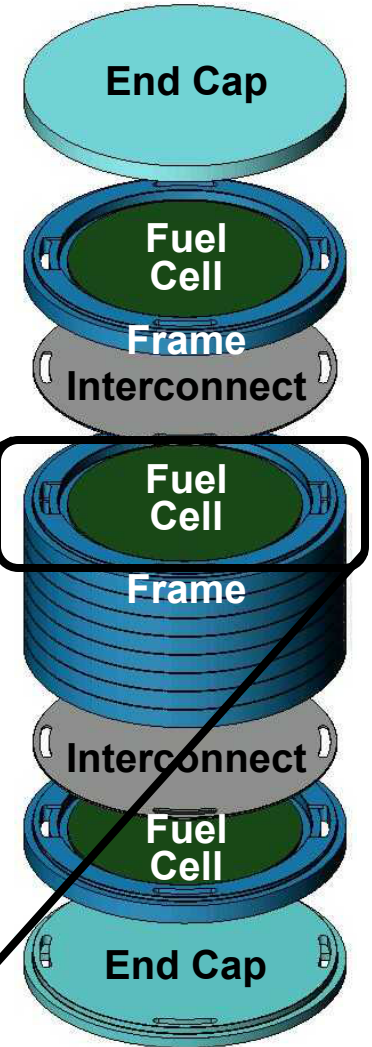


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- Operating temperature = 550 °C
 - Start-up challenges
 - Insulation materials
- CO₂ and H₂O reactants
 - Need preheating
 - Need pumps / blowers
- Methane – oxygen products
 - Must be stored at high pressure
- Dynamic control system

PCFC Stack Schematic



Initial studies on CO₂ hydrogenation and methane production shows reasonable selectivity to CH₄

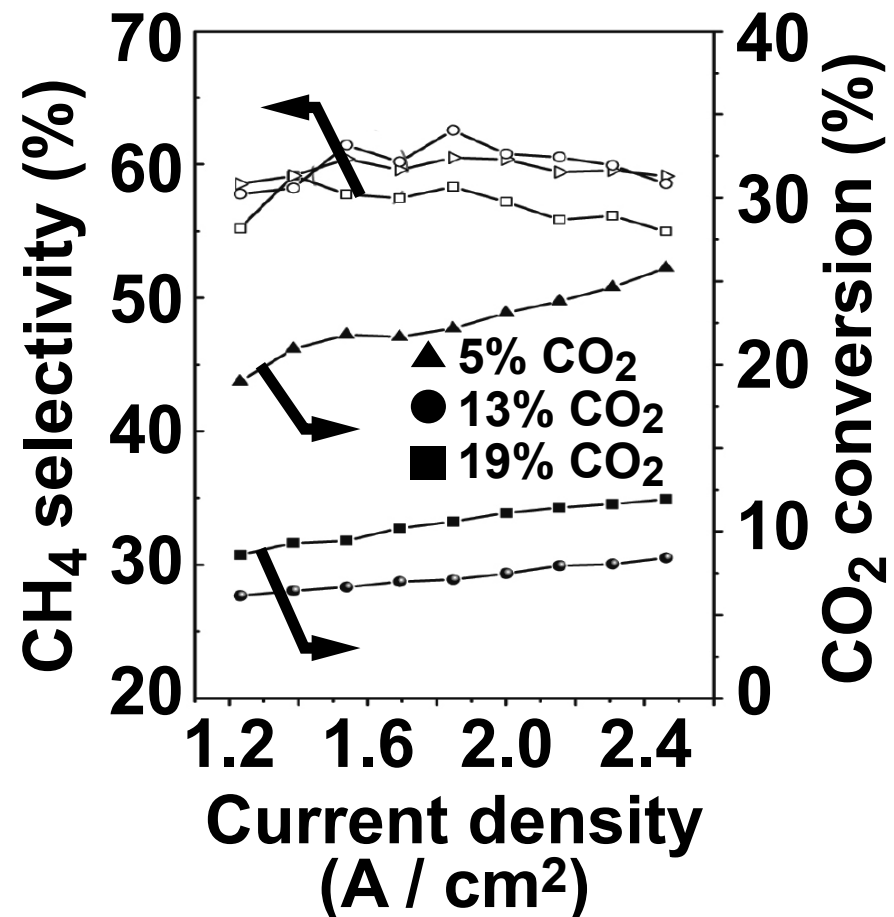


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- CO₂ conversion is modest
 - Generally < 25%
 - Necessitates large blowers
- CH₄ selectivity is encouraging
 - Approaches 60%
 - Independent of current density
- Increasing current density
 - Drives H⁺ across membrane
 - Promotes CH₄ formation
 - Requires more solar power

CO₂ conversion and CH₄ selectivity as a function of current density



In summary, protonic-conducting ceramics could play a role in manned missions to Mars



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- **Convert Martian-derived CO₂ into fuel and oxygen**
 - **Next-generation MOXIE**
- **Early stage of technological development**
 - **TRL 3-4?**
- **Development leveraged by terrestrial applications**
 - **Fuel cells for efficient electricity generation**
 - **Membrane reactors for renewable energy storage**
 - **Significant DOE support**
- **Could also serve as electric generator**
 - **Reversible operation is possible**
- **Industrial partner is moving technology to pre-commercial scale**
 - **FuelCell Energy, Danbury, CT**



Acknowledgements



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